

Quality of reclaimed water for turfgrass irrigation

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Cost for Increasing Water Supply

Option	Cost (acre-ft)
Desalination	\$1,400
Recycling for Drinking Water	\$1,230
Imperial Valley Imports	\$800
Current Costs	\$500
Landscape Reuse	\$70

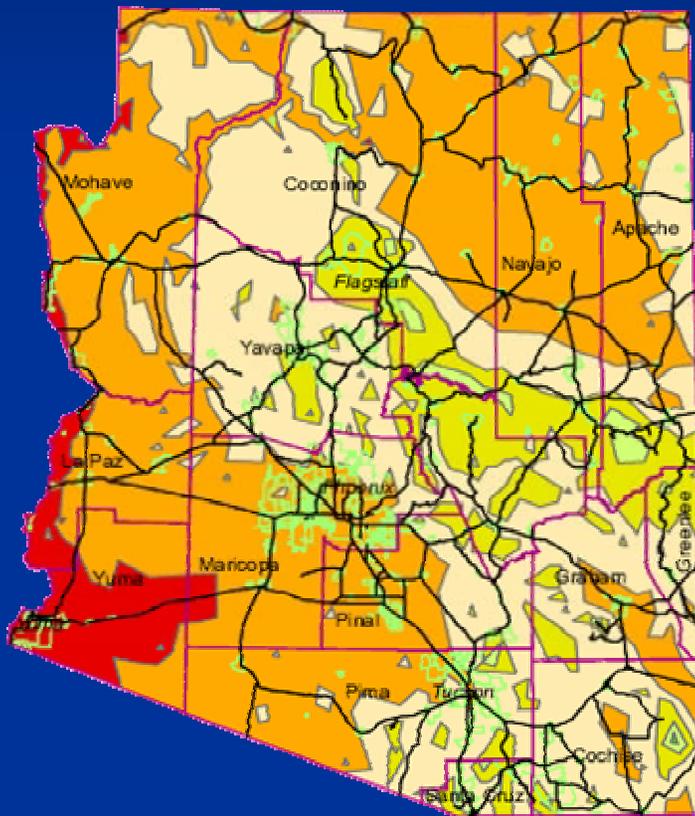


Water Quality Problems

- **Salinity**
 - Salts reduce water availability through osmotic effects
- **Water Infiltration Rate**
 - Relatively high sodium or low calcium content of soil or water reduces the infiltration rate
- **Specific Ion Toxicity**
 - Certain ions (sodium, chloride, or boron) from soil or water accumulate and cause damage
- **Miscellaneous**
 - Excessive corrosion of equipment

Salinity

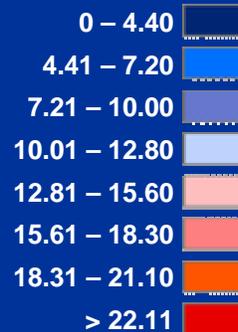
Precipitation



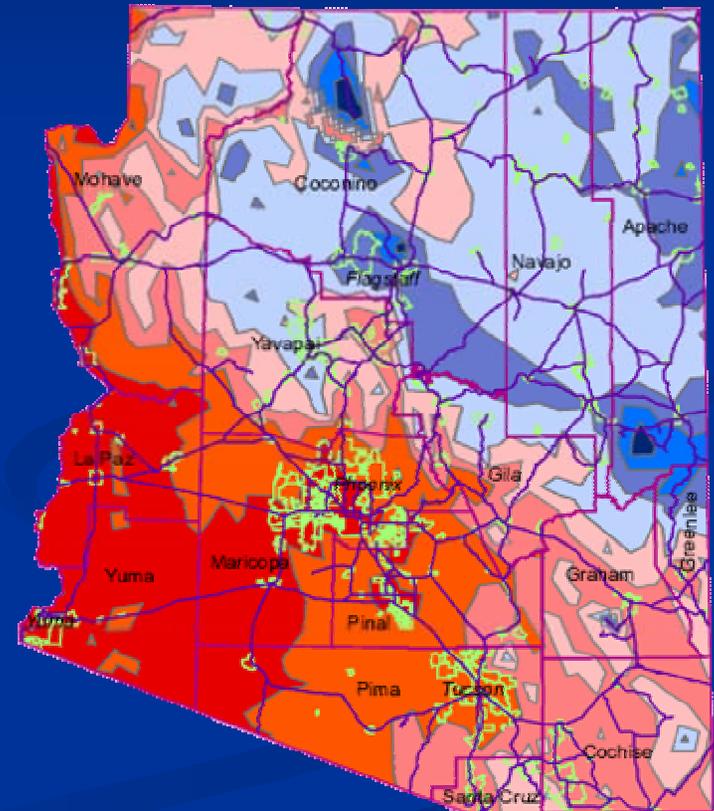
Precipitation (cm)



Temperature (C)



Temperature



Evaporative Demand > Precipitation

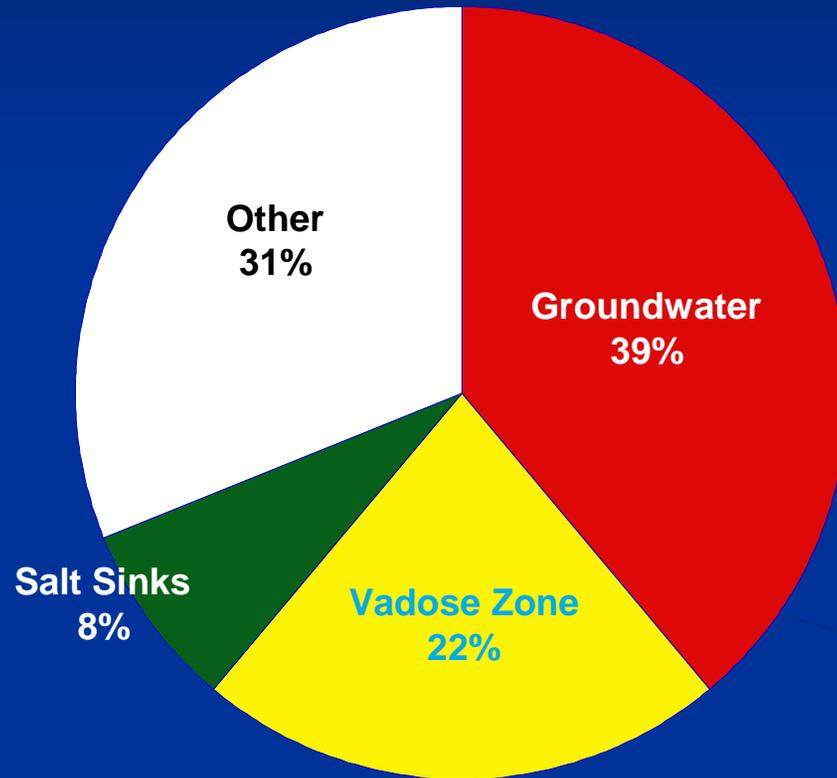
Salt River

- 0.8 maf/yr @ 500 mg L⁻¹
imports 620,000 metric tons
annually

Central Arizona Project

- 1.2 maf/yr @ 650 mg L⁻¹
imports 960,000 metric tons
annually

Salt Accumulation



Units for Salinity

- TDS – Total dissolved solids (mg L^{-1})
 - Measured by filtering then evaporating a sample and weighing the residue
 - Common measurement at sewer treatment plants
- EC – Electrical conductivity
 - Measured by passing a current through a water sample
 - Most management criteria based on EC

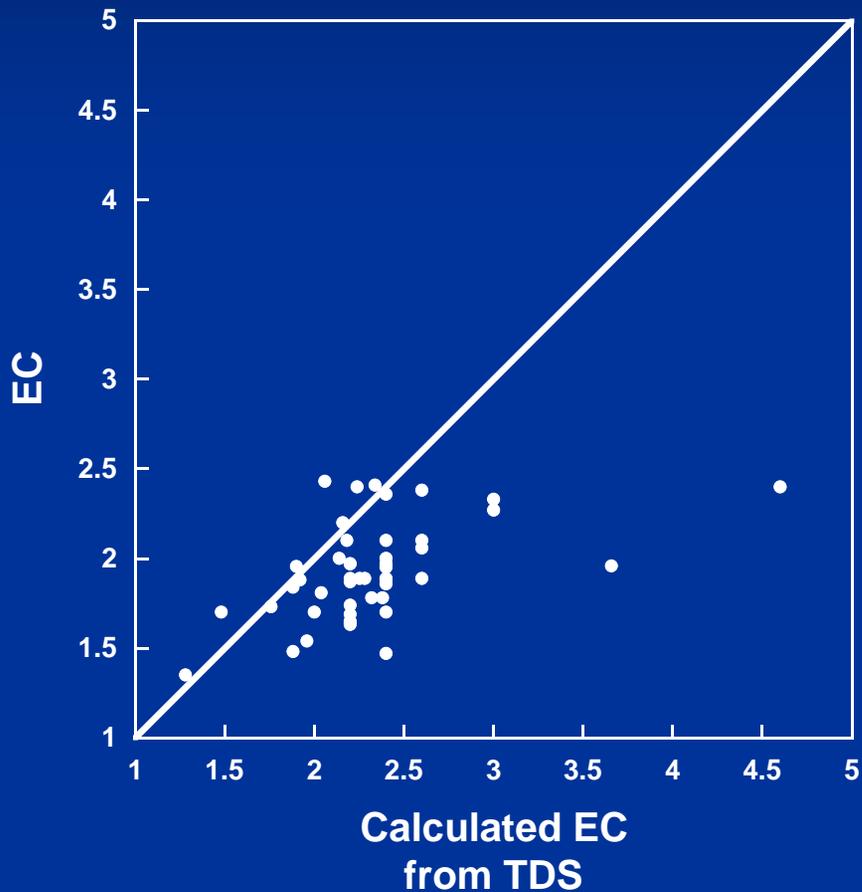
Converting between TDS and EC

$$\text{EC} \left(\frac{\text{dS}}{\text{m}} \right) = \frac{\text{TDS} \left(\frac{\text{mg}}{\text{L}} \right)}{\text{Conversion Factor}}$$

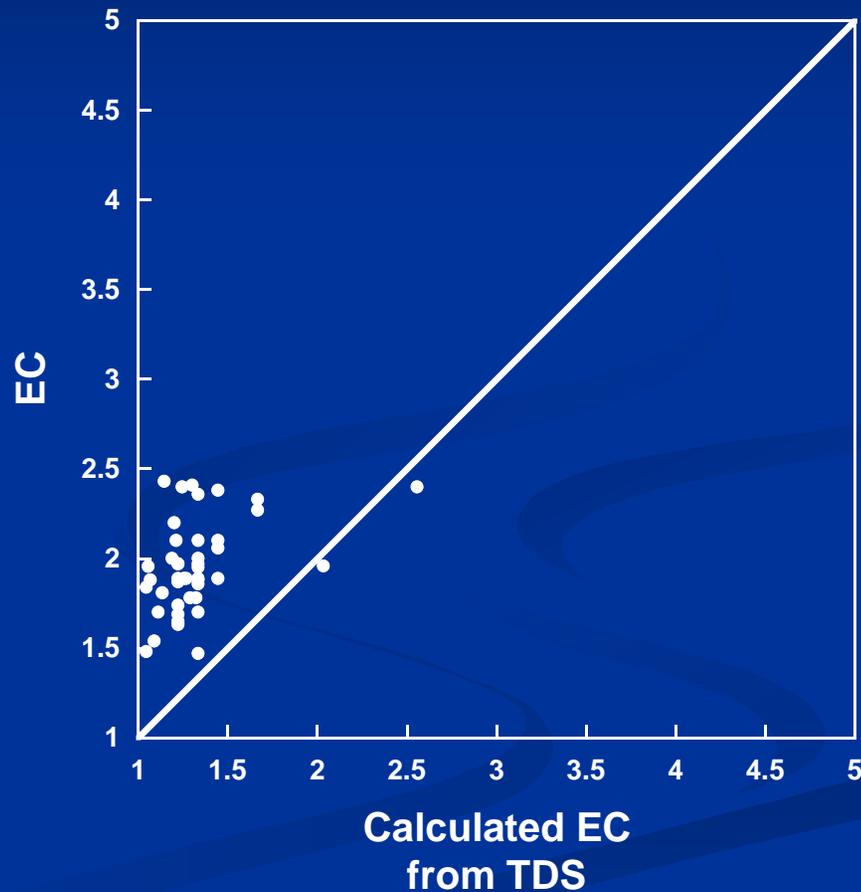
$$\text{TDS} \left(\frac{\text{mg}}{\text{L}} \right) = \text{EC} \left(\frac{\text{dS}}{\text{m}} \right) (\text{Conversion Factor})$$

Conversion factor usually 500 – 1000
depending on solution

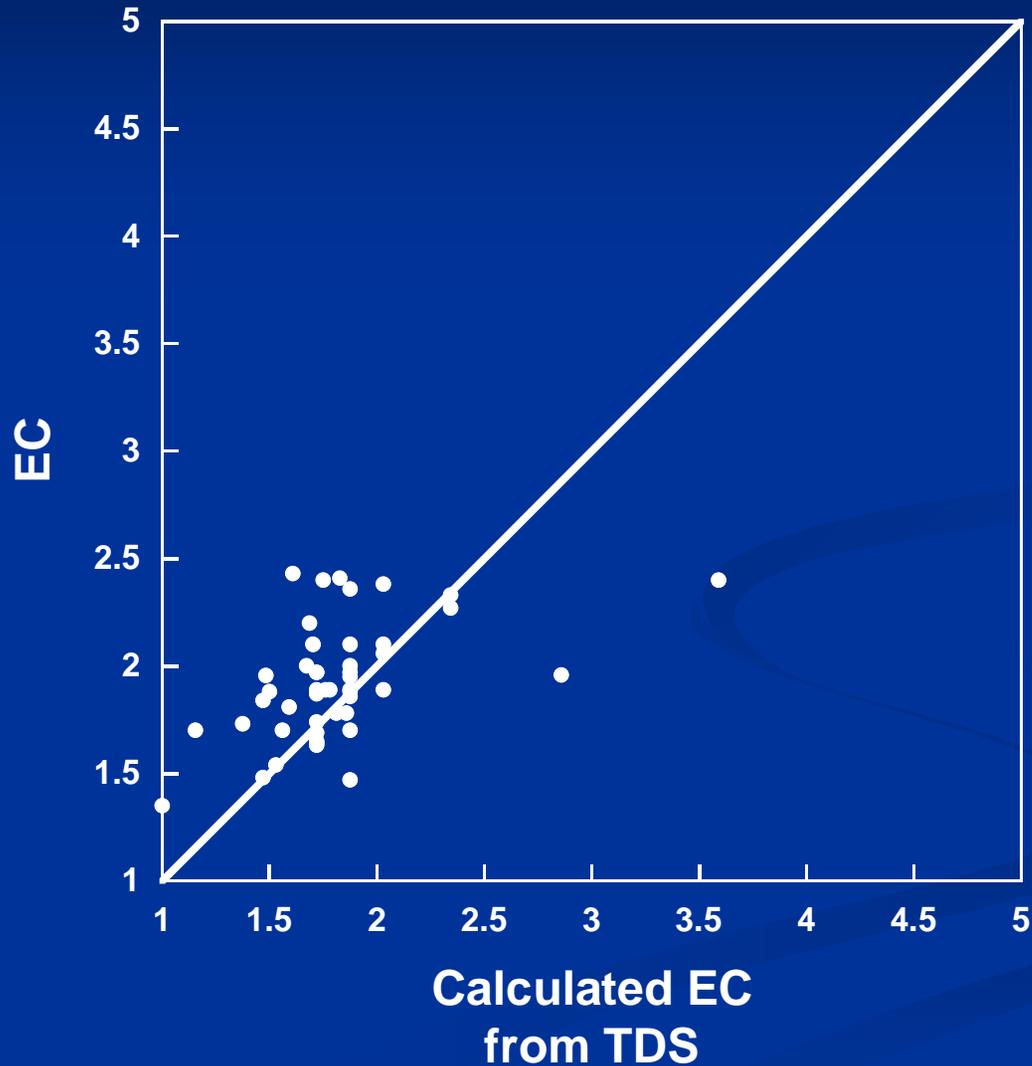
Conversion factor = 900



Conversion factor = 500



For most irrigation waters conversion factor ≈ 640



Units for EC

$$\frac{\text{dS}}{\text{m}}$$

$$\frac{\mu\text{S}}{\text{cm}}$$

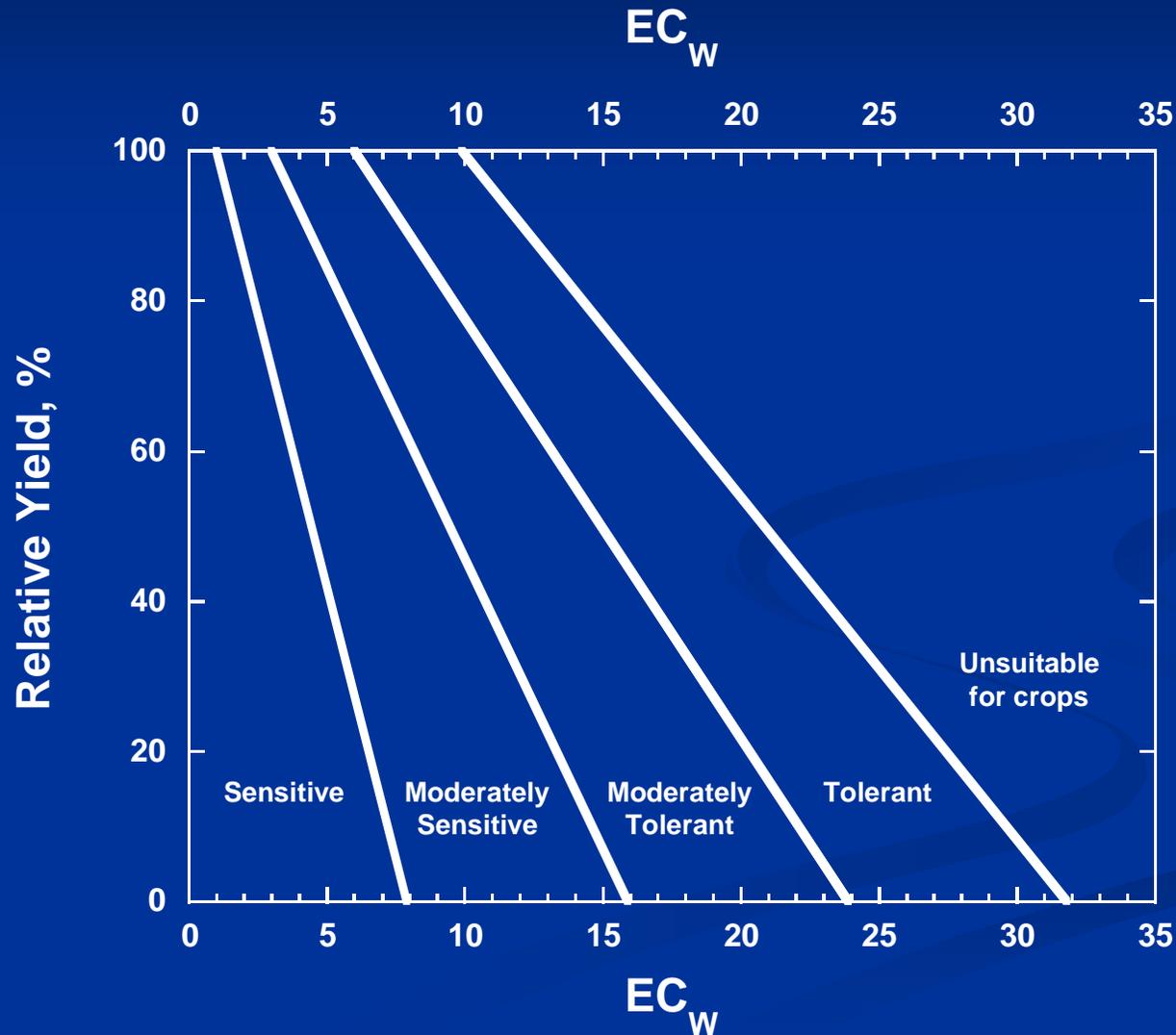
$$\frac{\text{mmho}}{\text{cm}}$$

$$\frac{\mu\text{mho}}{\text{cm}}$$

Conversion for EC units

$$\frac{\text{dS}}{\text{m}} = \frac{\mu\text{S}}{\text{cm}} = \frac{\text{mmho}}{\text{cm}} = \frac{\mu\text{mho}}{\text{cm}} \times 1000$$

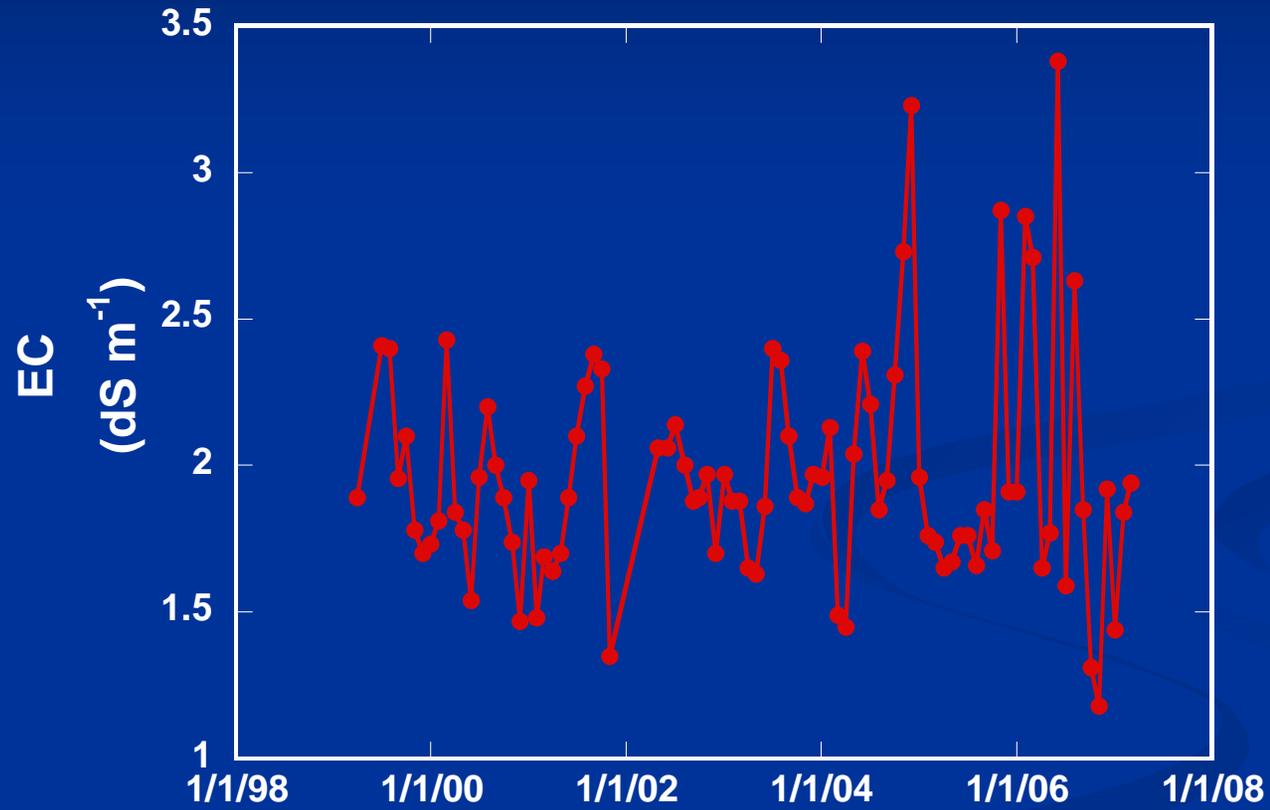
Irrigation with Saline Water



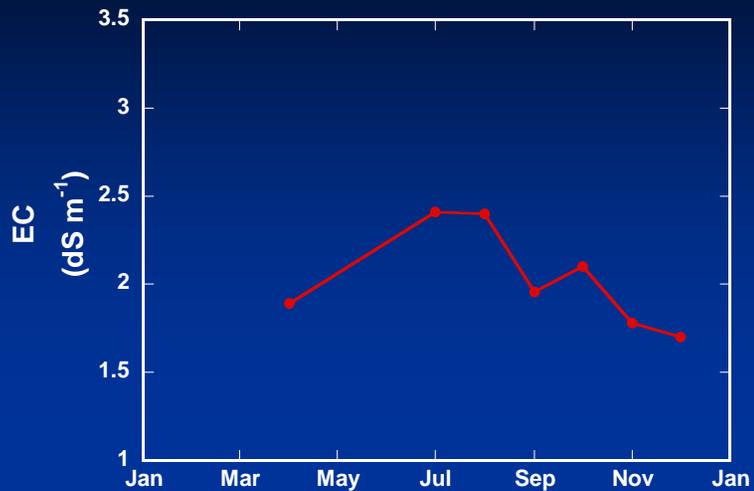
Irrigation Water Quality – Salinity

		Degree of Restriction on Use		
	Units	None	Slight to Moderate	Severe
EC_w	dS/m	< 0.7	0.7 – 3.0	> 3.0
(or)				
TDS	mg/l	< 450	450 – 2000	> 2000

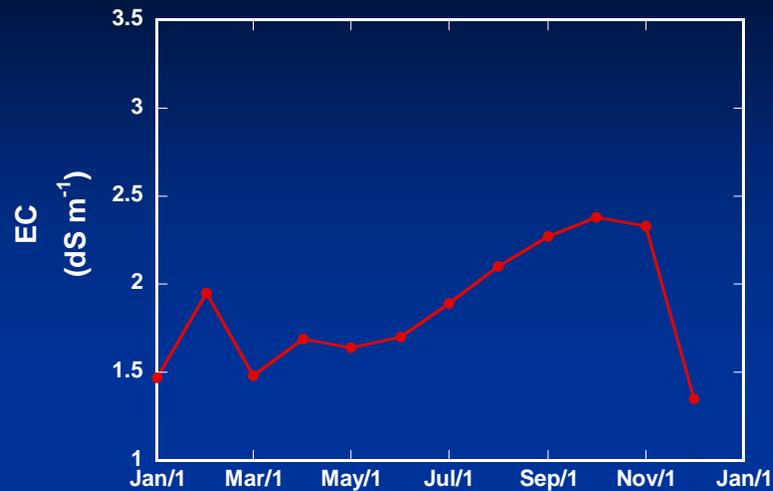
Chandler 1999-2007



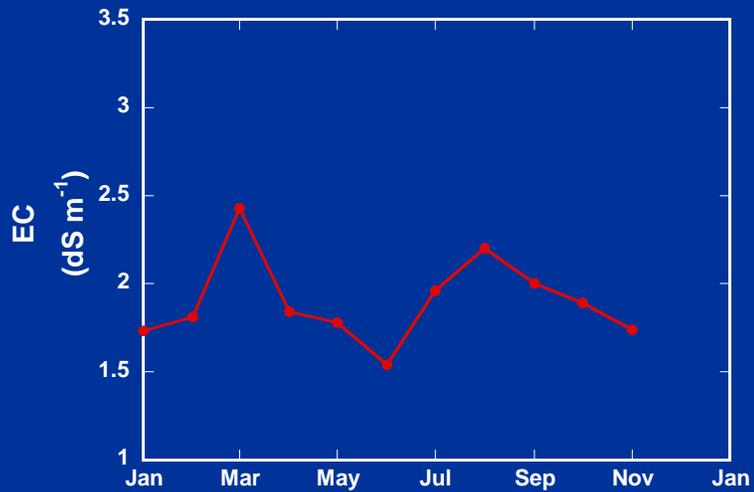
Chandler 1999



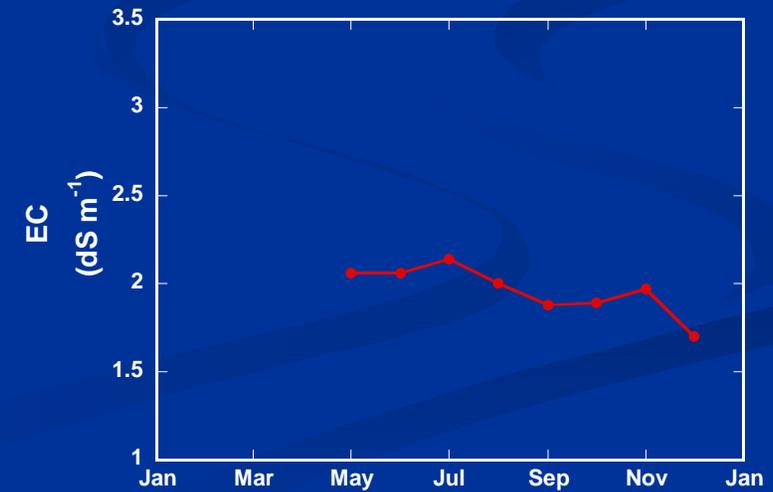
Chandler 2001



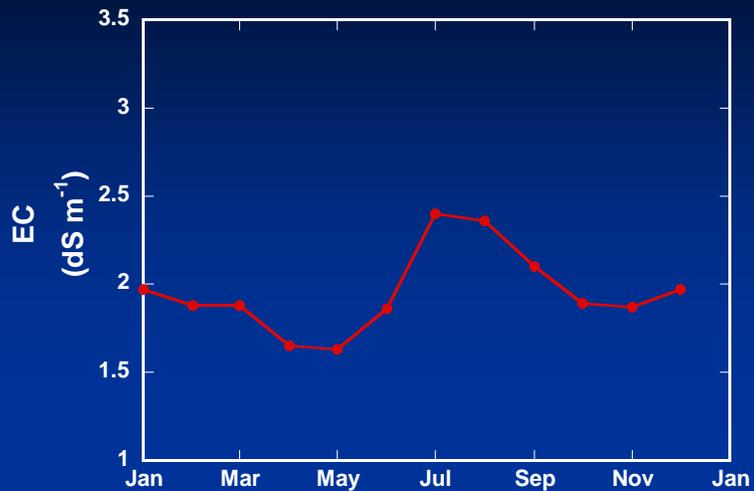
Chandler 2000



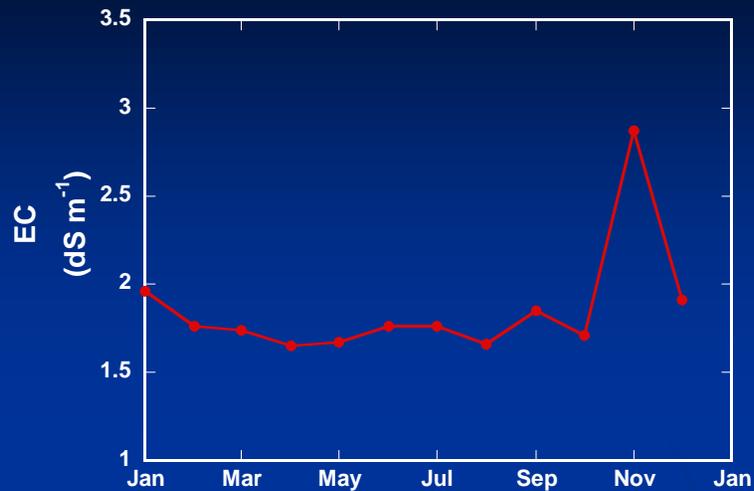
Chandler 2002



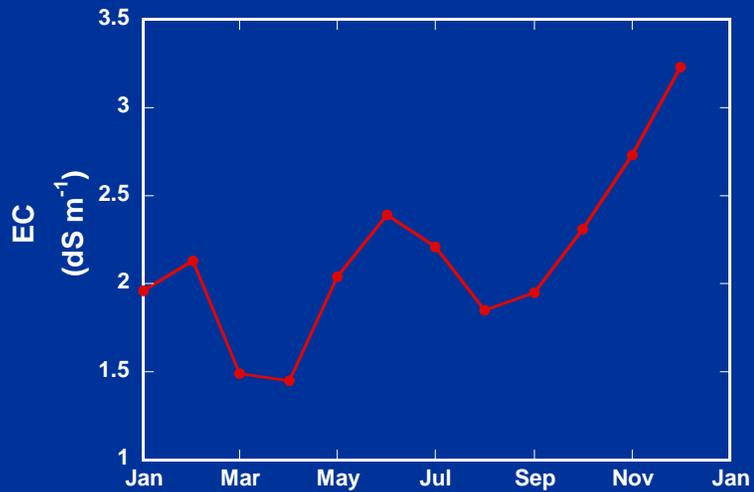
Chandler 2003



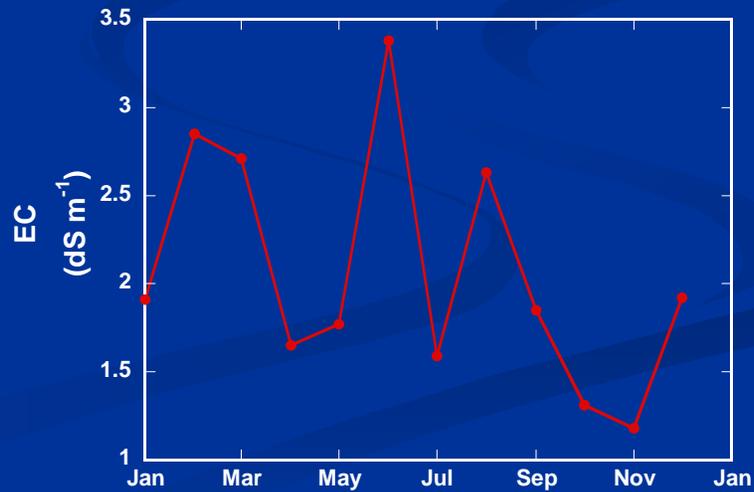
Chandler 2005



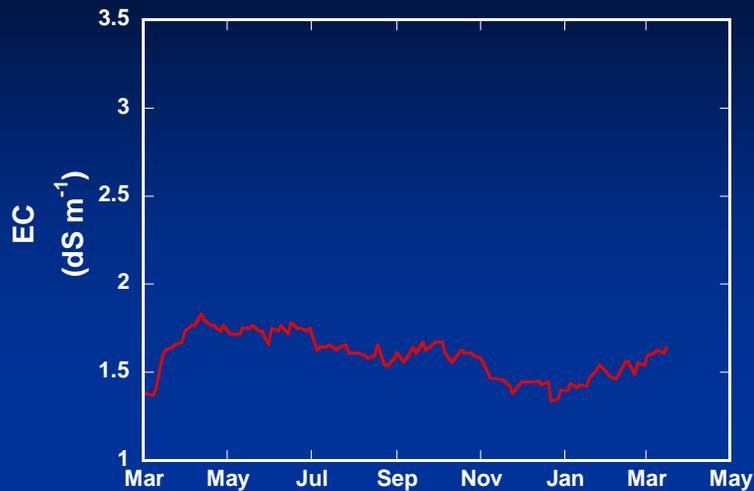
Chandler 2004



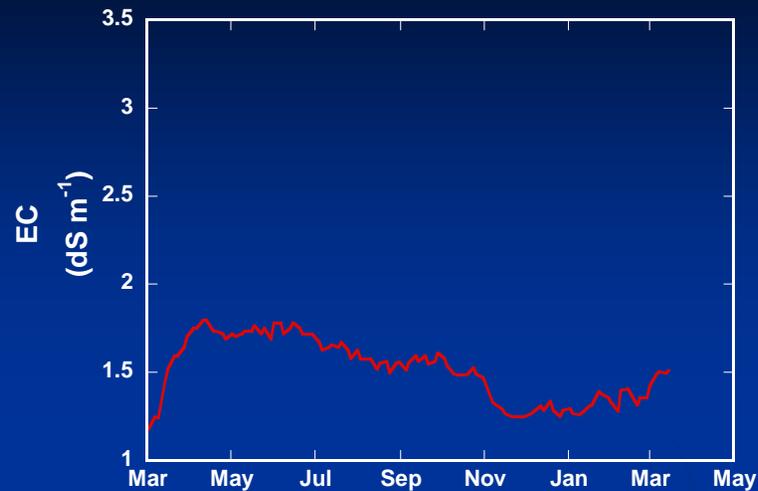
Chandler 2006



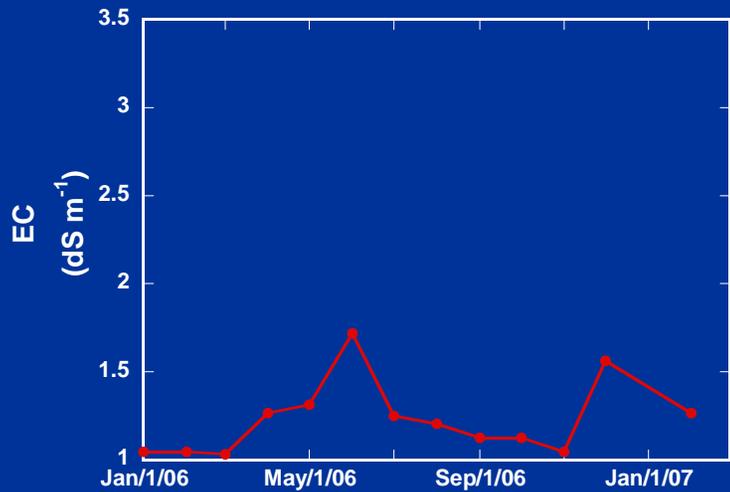
91st Avenue



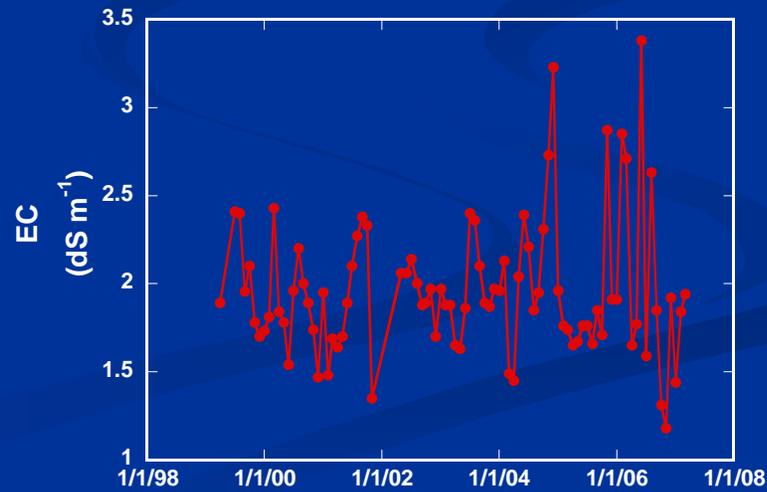
23rd Avenue



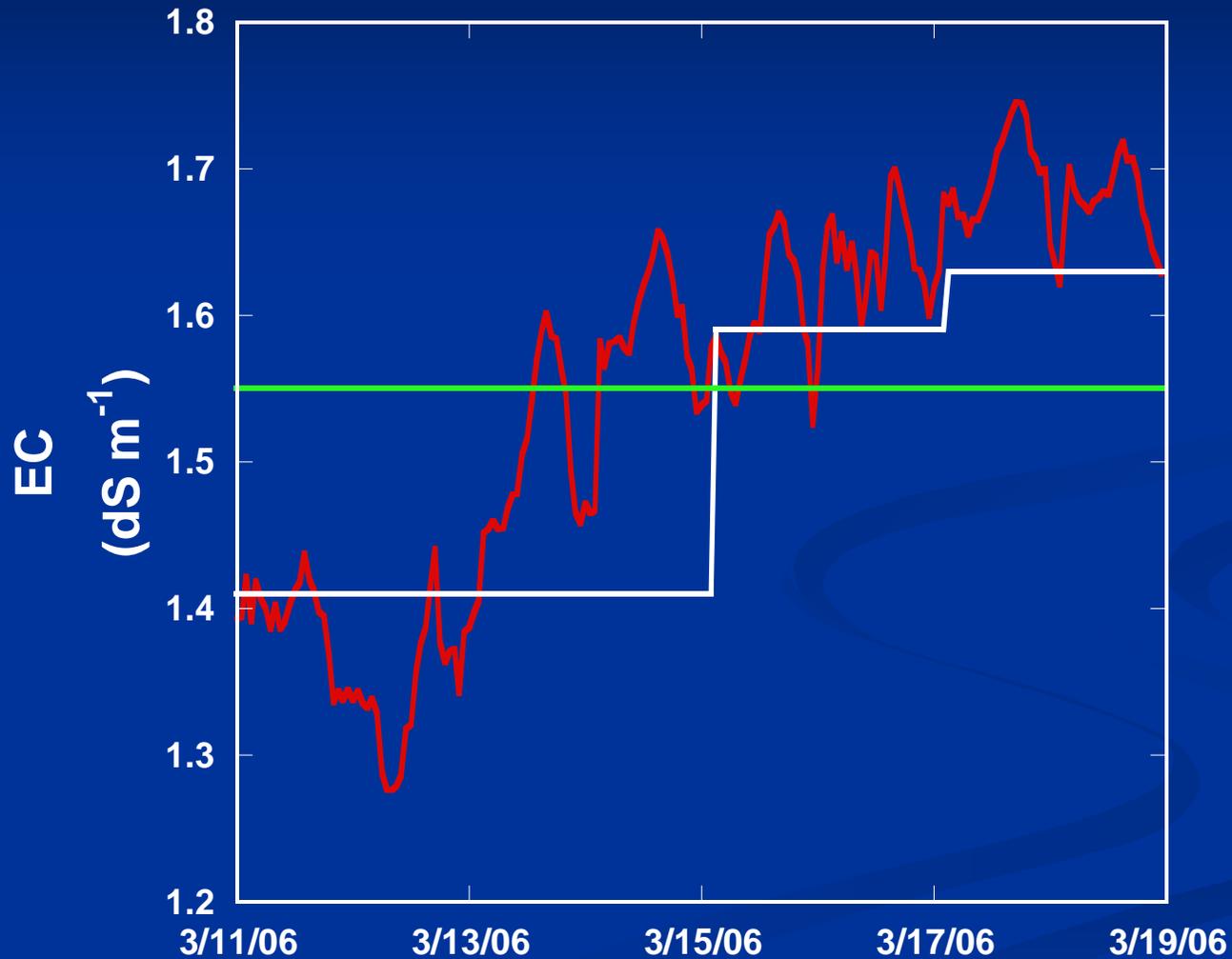
Gilbert 2006-07



Chandler 1999-2007



Data Collection Rate



Measure salinity often

- EC meters range in price from \$50 - \$1000
- Results are instant
- Track results



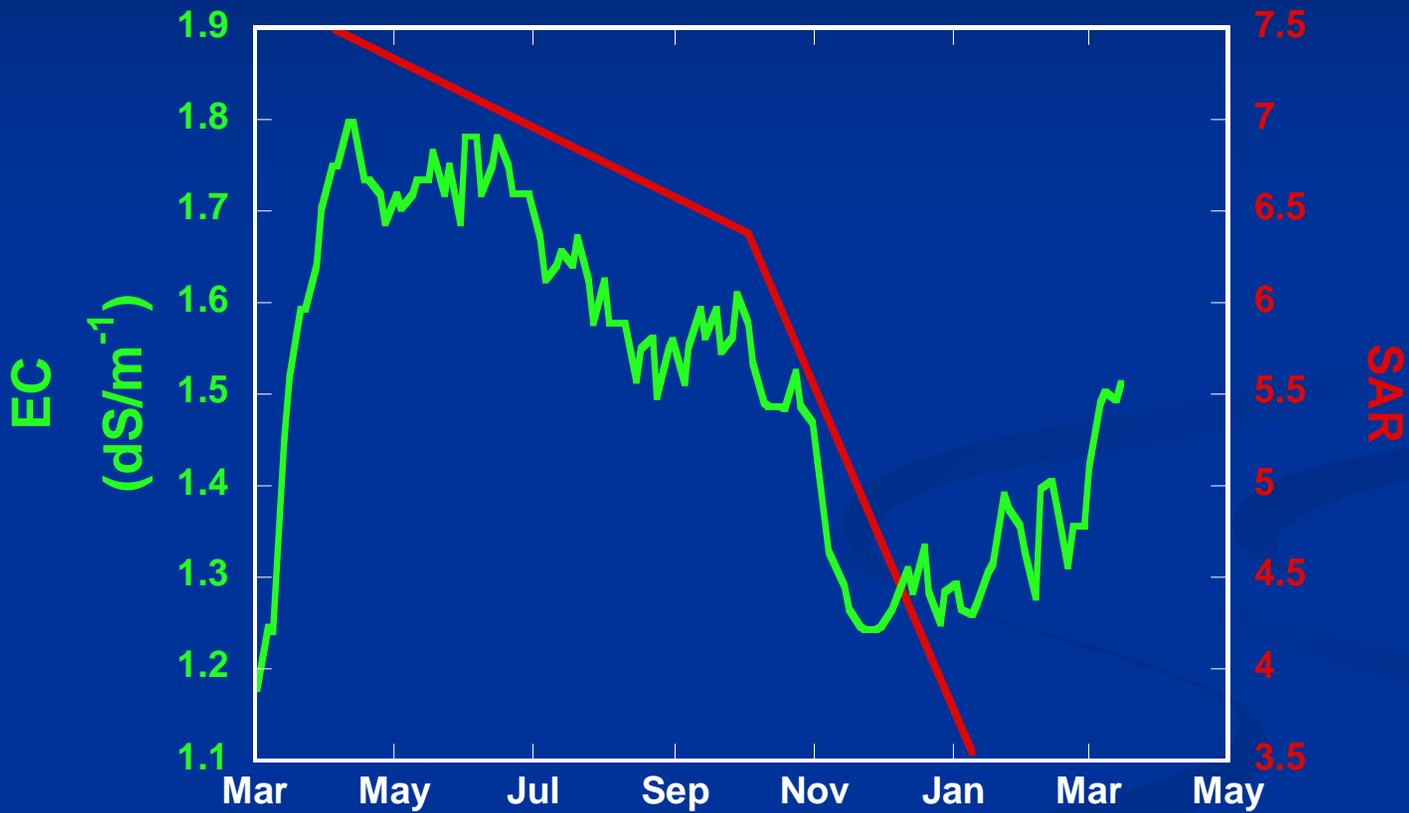
Irrigation Water Quality – Sodicty

Potential Infiltration Problems

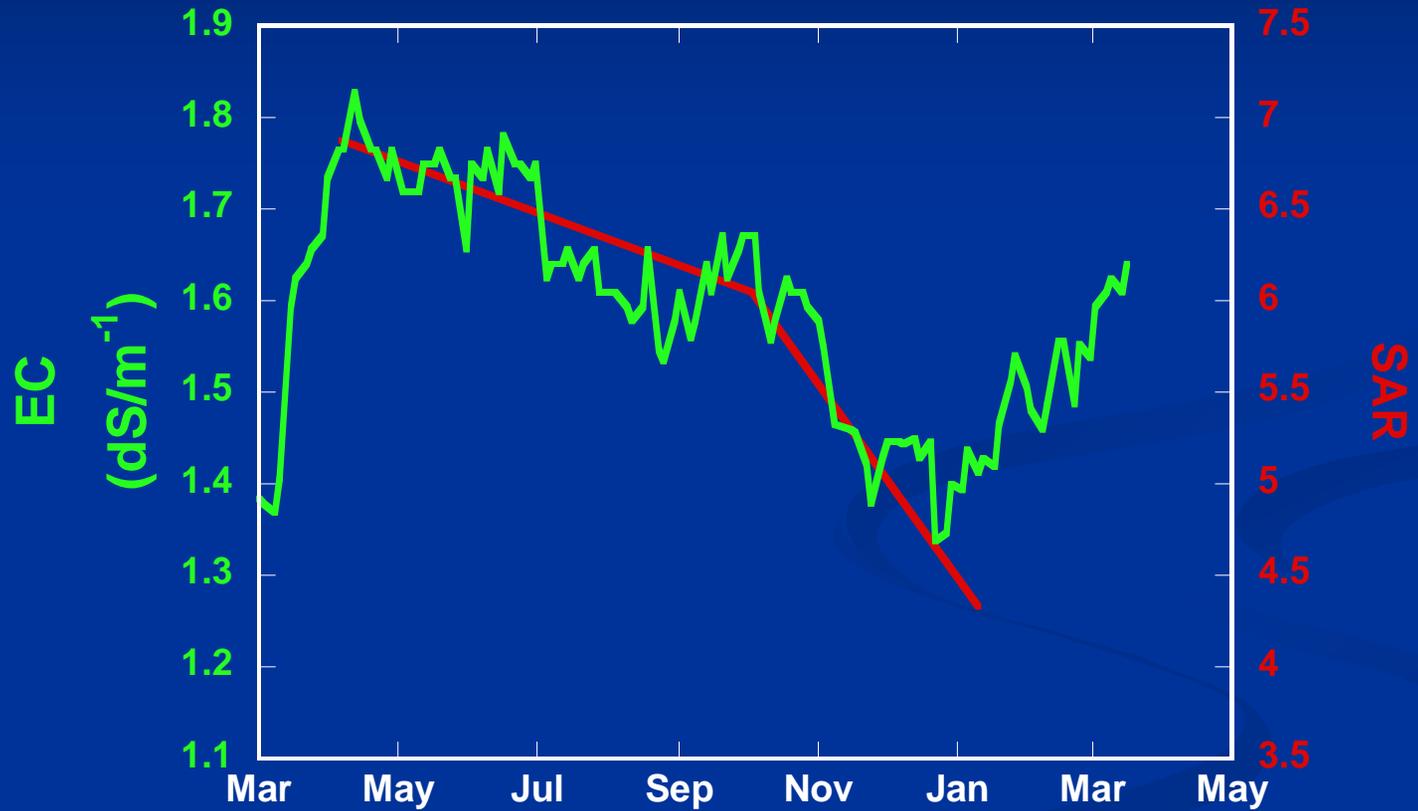
		Degree of Restriction on Use		
		None	Slight to Moderate	Severe
SAR = 0-3	and EC _w =	> 0.7	0.7 – 0.2	<0.2
= 3-6	and EC _w =	> 1.2	1.2 – 0.3	<0.3
= 6-12	and EC _w =	> 1.9	1.9 – 0.5	<0.5
= 12-20	and EC _w =	> 2.9	2.9 – 1.3	<1.3
= 20-40	and EC _w =	> 5.0	5.0 – 2.9	<2.9

$$\text{SAR} = \frac{[\text{Na}^+]}{\sqrt{\frac{[\text{Ca}^{2+} + \text{Mg}^{2+}]}{2}}}; \text{Conc} \Rightarrow \left[\frac{\text{meq}}{\text{l}} \right]$$

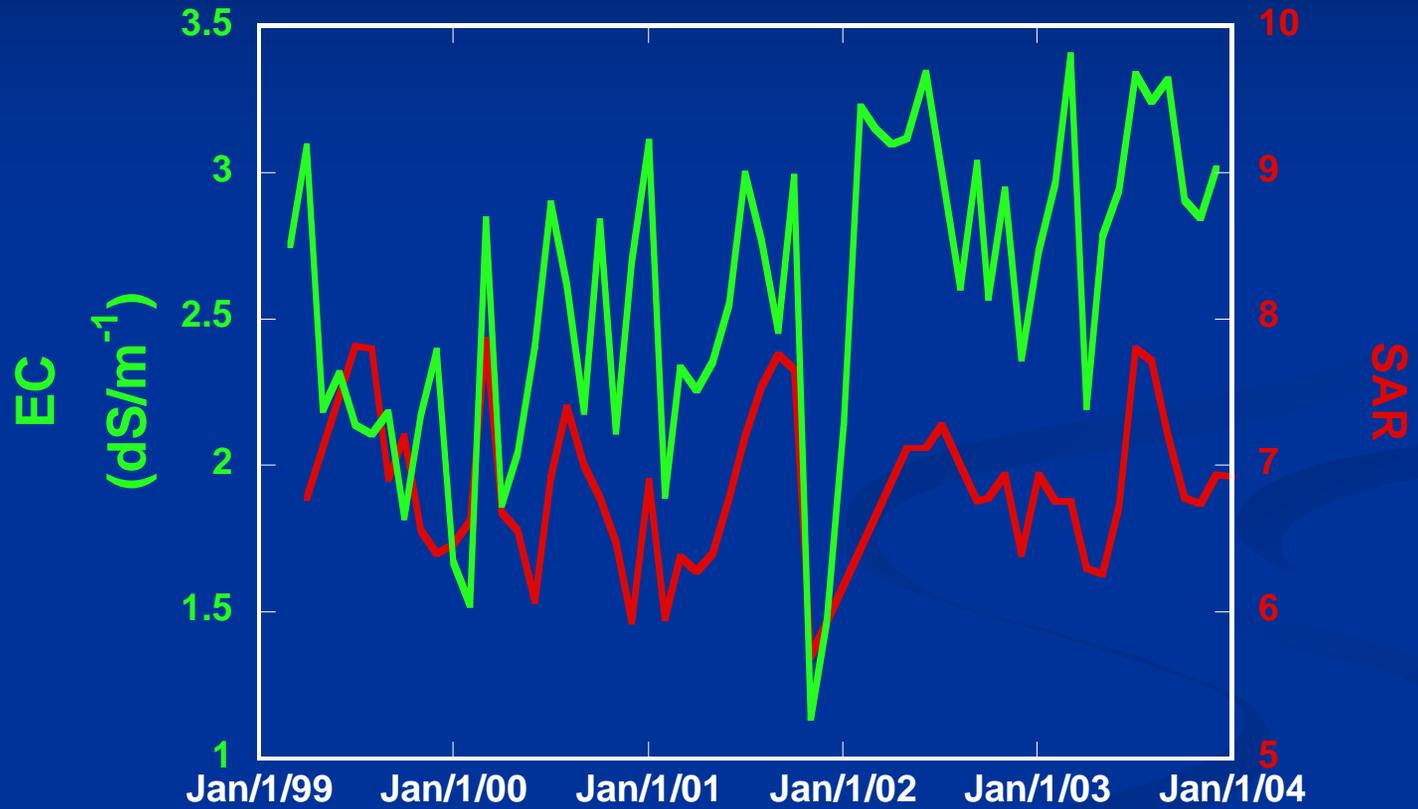
23rd Avenue



91st Avenue



Chandler



Irrigation Water Quality – Toxicity

Boron (ppm)

Class of Water	Tolerance		
	Sensitive	Semi-tolerant	Tolerant
Excellent	< 0.33	< 0.67	< 1.00
Good	0.33 to 0.67	0.67 to 1.33	1.00 to 2.00
Permissible	0.67 to 1.00	1.33 to 2.00	2.00 to 3.00
Doubtful	1.00 to 1.25	2.00 to 2.50	3.00 to 3.75
Unsuitable	> 1.25	> 2.50	> 3.75

Boron accumulates in leaf tips – removal in clippings

Problem for trees and shrubs

Salinity and wastewater reuse



- **Power plant has zero discharge permit for spent cooling water.**

Salinity and wastewater reuse



- **Waste water is used to irrigate 250 acre farm.**

Salinity and wastewater reuse



- Ongoing monitoring to determine if salt is leaching out of root zone and the effective life of the farm for waste water disposal.

Experimental Design



- **Two line source sprinkler systems.**
 - One waste water
 - One fresh water

Experimental Design



- **Six irrigation levels on either side of line source.**

Experimental Design



- **Water level 4 is approximately equal to no leaching beyond the root zone.**

Experimental Design

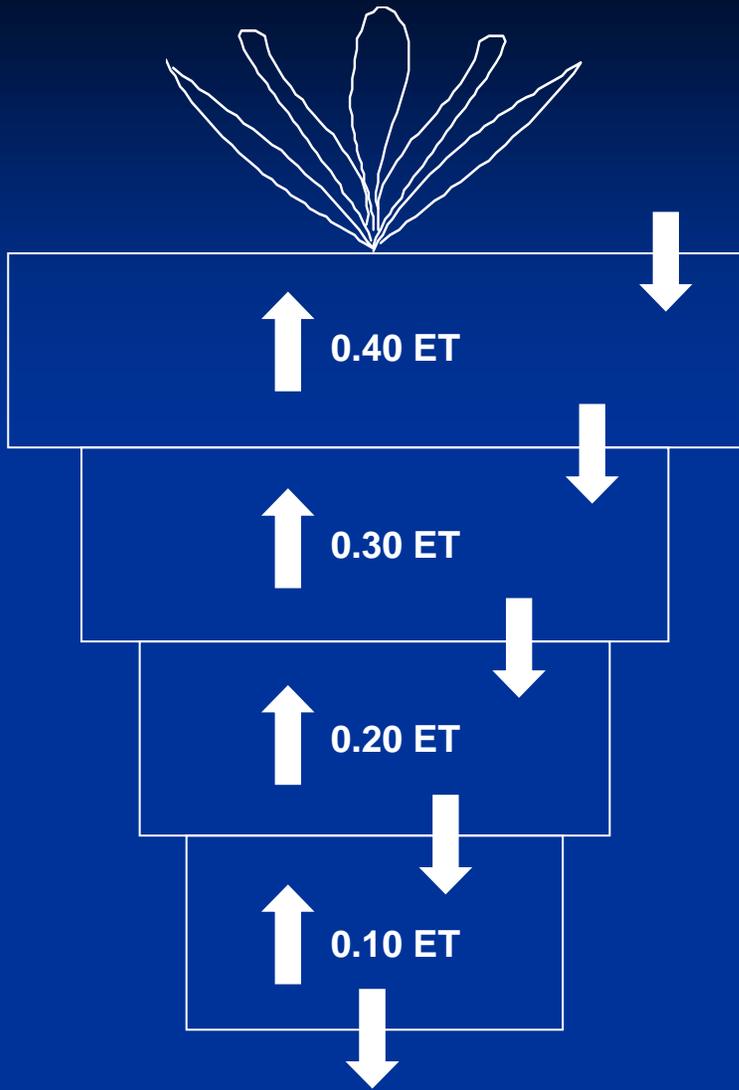


- **Water level 5 = LF 0.075**
- **Water level 6 = LF 0.15**

Properties of Waste Water

Concentration (mg L⁻¹)

Ca	410
Mg	158
Na	268
K	13.2
SO ₄	1629
Cl	276
B	317
EC	4.3 dS m ⁻¹



LF= 0.0066

LF= 0.075

LF= 0.150

EC₀ = 4.3 dS m⁻¹

EC₀ = 4.3 dS m⁻¹

EC₀ = 4.3 dS m⁻¹

LF₀ = 1.00

LF₀ = 1.00

LF₀ = 1.00

EC₁ = 7.1 dS m⁻¹

EC₁ = 6.83 dS m⁻¹

EC₁ = 6.5 dS m⁻¹

LF₁ = 0.602

LF₁ = 0.63

LF₁ = 0.66

EC₂ = 23.2 dS m⁻¹

EC₂ = 19.4 dS m⁻¹

EC₂ = 16.3 dS m⁻¹

LF₂ = 0.300

LF₂ = 0.353

LF₂ = 0.40

EC₃ = 221 dS m⁻¹

EC₃ = 116 dS m⁻¹

EC₃ = 71.0 dS m⁻¹

LF₃ = 0.110

LF₃ = 0.170

LF₃ = 0.23

EC₄ = 33,000 dS m⁻¹

EC₄ = 1,541 dS m⁻¹

EC₄ = 472 dS m⁻¹

LF₄ = 0.0066

LF₄ = 0.075

LF₄ = 0.15

ARZS

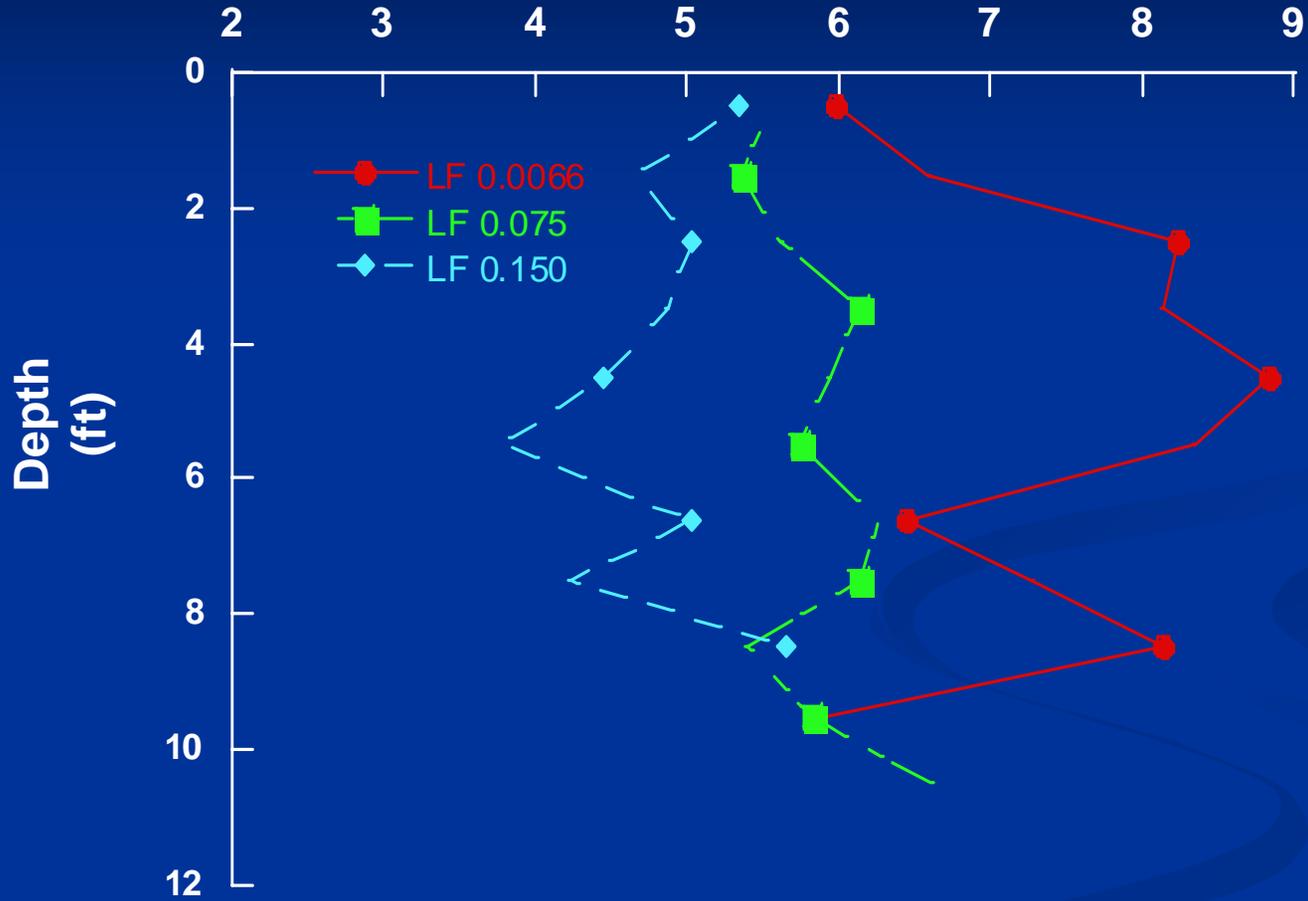
6650 dS m⁻¹

338 dS m⁻¹

114 dS m⁻¹

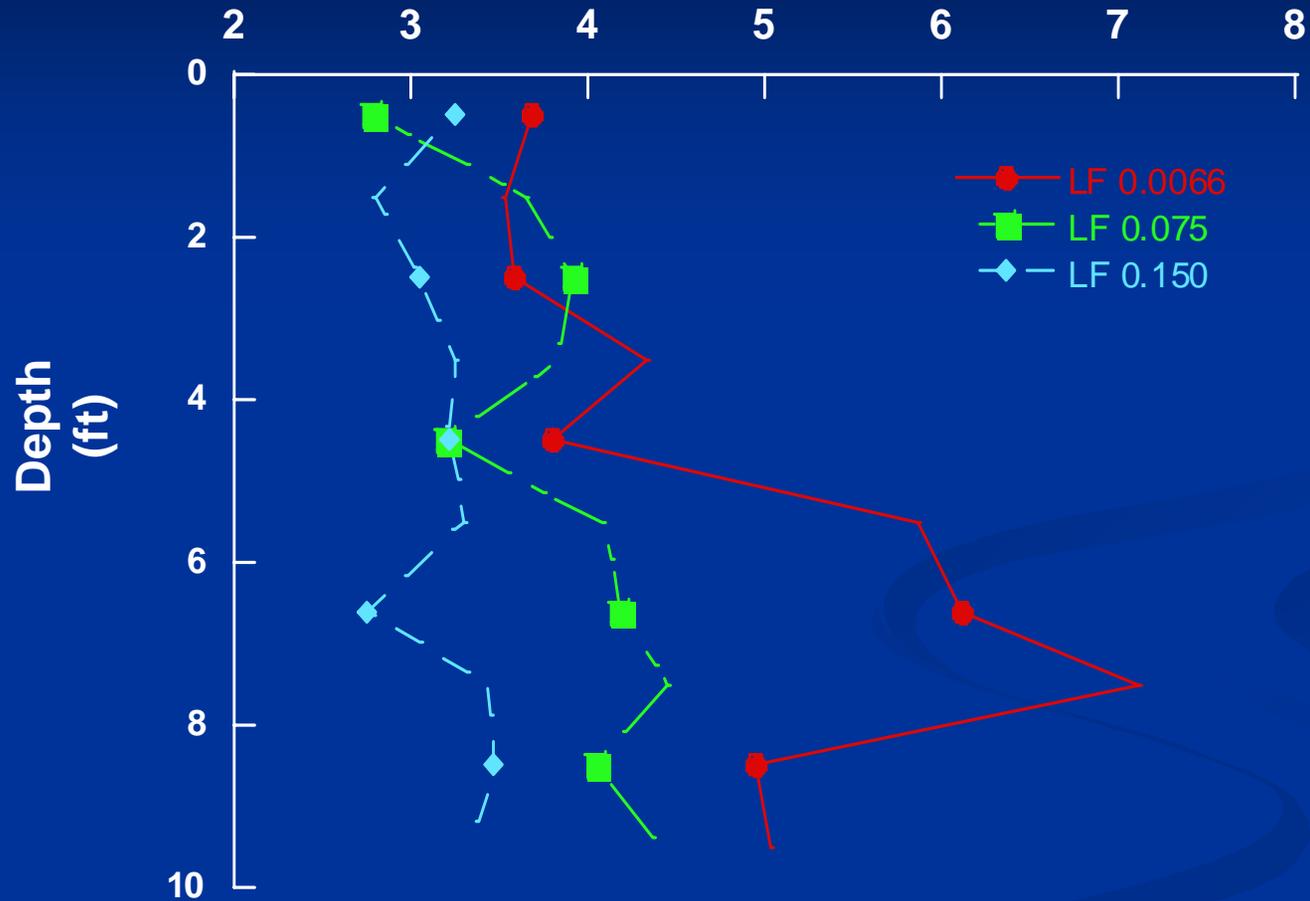
1999

Electrical Conductivity (dS m^{-1})



2000

Electrical Conductivity (dS m^{-1})



Results

- Salts in solution are not accumulating as fast as predicted.
- CaSO_4 precipitation.
- Potential B, Cl, and Na toxicity.

Managing Salinity and SAR

- Measure salinity often
- Monthly samples for SAR
 - Sodium, Calcium, Magnesium
- Quarterly sampling for toxicity
 - Boron, Sodium and Chloride
- Proper leaching fraction
- Replace sodium with polyvalent ions
(Ca^{2+})

Managing Salinity and SAR

- **Record Keeping**

